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TANTA U NIVERSITY FACULTY OF SCIENCE DEPARTMENT OF GEOLOGY EXAMINATION FOR SECOND LVEL STUDENTS OF SPECIAL GEOLOGY COURSE TITLE: DATE: JAN. 2017 TERM: FIRST TOTAL ASSESSMENT MARKS: 100 TIME ALLOWED: 2 HOURS

Answer the following questions: (60 Marks)

- 1- Write short article on composition, colour, classification and geological record of amber.
- 2- Write short article on physical properties, origin and treatment of turquoise.
- 3- Gem varieties of beryl.
- 4- Physical properties and gem varieties of tourmaline.
- 5- Physical properties and origin of diamond.
- 6- Write briefly on the following: (40 Marks)
- a- Treatment of gemstones.
- b- Colour and varieties of topaz.
- c- Jet and Momme weight of pearl.
- d- Major varieties of crystalline silica.
- e- Physical properties of lazurite and enhancement of jade.

Examiner: Prof. Ibrahim Salem

وحدة ضمان الجودة (١٥) كلية العلوم - جامعة طنطا (١٥) U CUALITY ASSURANCE UNIT

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TANTA UNIVERSITY FACULTY OF SCIENCE DEPARTMENT OF GEOLOGY

EXAMINATION FOR (LEVEL 2) SPECIAL GEOLOGY, CHEMISTRY/GEOLOGY AND GEOPHYSICS

1910	COURSE TITLE:	Principles of St	ratigraphy	CODE: GE 2107
DATE:	JANUARY, 2017	SEMESTER: 1	TOTAL MARKS:100	TIME ALLOWED: 2 HOURS

Answer the following questions (Illustrate your answer with drawing):

Question 1: Discus briefly the subsurface stratigraphic procedures.	(20 Marks)
Question 2: State and explain the law of Faunal Succession and the Inclusion Princip	(20 Marks) le.
Question 3: Write briefly about: a - Maastrichtian Age. b - Angular unconformity.	(20 Marks)
Question 4: Complete the following: a- Phanerozoic is subdivided into	smallest ones, not the
Question 5:	(20 Marks)

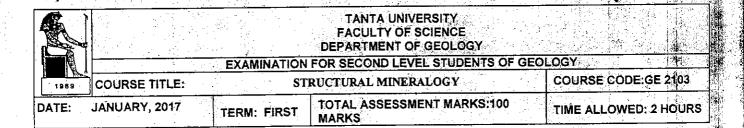
Match the number of the term or concepts in Column 1 with the letter of the correct definition in Column 2.

1- Precambrian	a- Using fossils to determine age	
2- Absolute age	b- Sediments on igneous or metamorphic rocks	
3- Uniformitarianism	c- Last occurrence of a species	
4- Bed	d- Fining Upwards Sequence	
5- Extinction	e- Present is the key to the past	
.6- Relative dating	f- Based on half lives of radioactive decay	
7- Regression	g. Older than 560 million years	
8- Transgression	h. Basic unit of stratigraphy	
9- Nonconformity	i- Epoch do we live in	
10- Holocene	j- More landward facies overlie more basin-ward facies	

Best wishes

		<u> </u>
Examiners	Prof. Dr. H. Khalil	D. M. C.LL
DAUMING	riot. Dr. H. Kilaili	Dr. M. Sobhy
	L	J

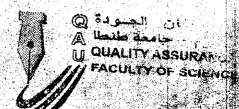
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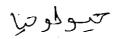


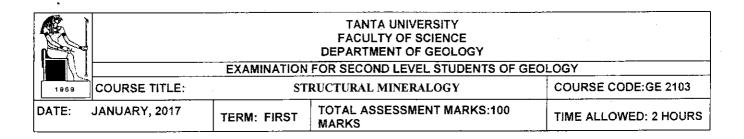
1.1	.Write short notes on the followings. Illustrate your answers with diagrai	ns ,	whenever
	ossible:		
-		(60	mark)
	Calcite structure.	Meson Nett	
b.	Unit cell of monoclinic system.		
c.	The structure of gold.		
	Cuprite structure.		
	Two dimensional lattice planes.		
f.	Sorosilicate structure.		
·			
II]	Define	(20) mark)
a	Screw axis		
	Coordination number		
	Isomorphism		
d.	. Electrostatic valency strength		
e.	Mesodesmic		
Ш	II Complete the followings:	(2	0 mark)
•. •			Ne.
a.	Two is the number of oxygen shared in		il
b.	is the smallesz possible subdivision which has the properties	01	the vision
	macrocrystal.	18 g	
	Ionic minerals are classified into		
u.	Si · O ratio in * equal 12		

GOOD LUCK

EXAMINERS	PROF. SAMIR M, ALY	Prof. ABD El SALAM, M. ABU EL ELA
	PROF. GAAFAR A. EL BAHARIYA	PROF, MOHAMED M. ABU ANBAR







I .Write short notes on the followings. Illustrate your answers with diagrams whenever possible:

(60 mark)

- a. Calcite structure.
- b. Unit cell of monoclinic system.
- c. The structure of gold.
- d. Cuprite structure.
- e. Two dimensional lattice planes.
- f. Sorosilicate structure.

II Define

(20 mark)

- a. Screw axis
- b. Coordination number
- c. Isomorphism
- d. Electrostatic valency strength
- e. Mesodesmic

III Complete the followings:

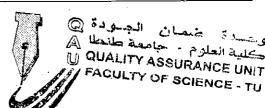
(20 mark)

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a.	Two is the number of oxygen shared i	n	•••••			
b.	is the smallesz poss	ible subdivision	which	has the	properties	of the
	macrocrystal.		a ·			
c.	Ionic minerals are classified into					
d.	Types of pseudomorphism are	******				•
e.	Si: O ratio inequ	al 1:2				

GOOD LUCK

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	Tanta University		
	Faculty of Science		
	Chemistry Department		
Examination for Second Year Students of Special Geology Section			
	Organic Chemistry	Course Code: CH2145	
	<u> </u>	Time Allowed: 2 hrs	
	Examination for S Course Title January 2017	Faculty of Science Chemistry Department Examination for Second Year Students of Specia Course Title Organic Chemistry	

Answer the following questions:

1) Differentiate between each of the followings:

(20 Mark)

- 1. Action of HCl and O₃ on 1-butene and 2-butene
- 2. Preparation of alkenes and alkynes using vicinal dihalide
- 3. Addition of water (H₂O) on ethyne and propyne
- 4. Structural and functional isomerism
- 5. The reaction of Grignard reagent with water and aldehydes & ketones

2) Write the mechanism of:

(15 Mark)

- 1. Action of Chlorine (Cl2) on methane
- 2. Nitration and Acylation of benzene
- 3. Addition of HCl on propene in the presence and in the absence of $H_2\mathrm{O}_2$

3) Discuss the followings:

(15 Mark)

- 1. Functional groups
- 2. Wurtz coupling reaction
- 3. Transference of 2° or 3° alcohols to 1° alcohols
- 4. ortho, meta, and para directing groups

With best wishes

EXAMINER	Prof. Dr. Adel selim	Dr. Mohamed Azaam



TANTA UNIVERSITY FACULTY OF SCIENCE DEPARTMENT OF GEOLOGY

EXAMINATION FOR SOPHOMORES (SECOND YEAR) STUDENTS OF GEOLOGY

DATE: 24 JAN. 2017 FIRST SEMESTER TOTAL ASSESSMENT MARKS: 100 TIME ALLOWED: 2 HRS

Answer the following questions:-

(30 marks)

- a. Cubic system.
- b. Orthorhombic system.
- c. Hexagonal system.
- d. Tetragonal system.
- e. Monoclinic system.
- 2- Classify the following crystallographic forms illustrating your answer with both clinographic and stereographic projections:- (40 marks)
 - a. Octahedron.
 - b. Hexagonal prism 1st order.
 - c. (A) dome.
 - d. Tetragonal bipyramid 2nd order.
- 3- Plot the stereographic projection for the following parameter faces:- (30 marks)

a. (101)

b. (011)

c. $(11\overline{1})$

d. (110)

e. $(22\overline{1}0)$

f. (1320)

Best Wishes

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Tanta University Faculty of Science Chemistry Department

Examination for Second Year Students of Special Geology Section

Course Title Organic Chemistry Course Code: CH2145

: January 2017

Total Assessment Marks: 50

Time Allowed: 2 hrs

Answer the following questions:

1) Differentiate between each of the followings:

(20 Mark)

- 1. Action of HCl and O₃ on 1-butene and 2-butene
- 2. Preparation of alkenes and alkynes using vicinal dihalide
- 3. Addition of water (H₂O) on ethyne and propyne
- 4. Structural and functional isomerism
- 5. The reaction of Grignard reagent with water and aldehydes & ketones

2) Write the mechanism of:

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- 1. Action of Chlorine (Cl2) on methane
- Nitration and Acylation of benzene
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3) Discuss the followings:

(15 Mark)

- 1. Functional groups
- 2. Wurtz coupling reaction
- 3. Transference of 2° or 3° alcohols to 1° alcohols
- 4. ortho, meta, and para directing groups

With best wishes

EXAMINER	Prof. Dr. Adel selim	Dr. Mohamed Azaam

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DATE:

DEPARTMENT OF GEOLOGY

EXAMINATION For the Second Level of Chemistry-Geology

Students

COURSE TITLE 22/1/2017

Optical Mineralogy

COURSE CODE: 2105

TOTAL ASSESSMENT MARKS:100

TIME ALLOWED: 2 hrs.

Answer the following questions. Illustrate your answer.

1. Explain WHY and/or HOW?

(30 marks)

a. Absent of 2V angle in uniaxial interference figures.

Final Exam

- b. Interference figure can be differentiated between isotropic minerals and pseudoisotropic ones.
- c. Transparent calcite rhomb displays double refraction.
- d. Intereference figures display intensity of bireferengce.
- e. Pleochroism is absent seen in some colored minerals.
- 2. Refractive indices display an important role in almost optical properties of the minerals. Explain how and why? (20 marks)
- 3. State whether the following statements are True of False and correct the false one?

(30 marks)

- a. Optic axis is obtained' by sections cut normal to the C-axis of uniaxial minerals.
- b. The parallel sections of tetragonal minerals give the lowest optical properties.
- c. Isogyres in interference figures are formed due to extinction position:
- d. All uniaxial interference figures have the same optical behavior, at rotation of the microscopic stage.
- e. Anomalous interference colors is most common property in isotropic minerals.
- f. Zoning is characteristic optical feature of quartz and biotite.
- 4. Sketch and label three types of uniaxial interference figures showing their relations to its indicatrix. (20 marks)

Wishing Success for the ALL

Examiner: Prof. Mohamed Th. S. Heikal

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DATE:

TANTA UNIVERSITY **FACULTY OF SCIENCE** DEPARTMENT OF GEOLOGY

EXAMINATION FOR SECOND LEVEL STUDENTS OF (GEOLOGY) - (GEOPHYSICS) -

(GEOLOGY-CHEMISTRY) COURSE TITLE: MICROPALEONTOLOGY (1) COURSE CODE: GE 2109 JANUAR, 2016 TERM: FIRST TOTAL ASSESSMENT MARKS: 100 TIME ALLOWED: 2 HOURS

1) Write short notes on Five of the following questions. Illustrate your answers with clear drawings and give examples:

A.	Mixed chambers arrangement of test.	(Five only)	(10 Marks)
В.	Shape of the apertures in Foraminifera	(<u>Five only</u>)	(10 Marks)
C.	Mode of coiling in foraminifera		(10 Marks)
D. Sutures in Foraminifera.			(10 Marks)
Ĕ,	Dimorphism in Foraminifera.		(10 Marks)
F.	Application of Foraminifera		(10 Marks)

2) Give Examples: (15 Marks)

- A. Biumbonate test.
- B. Surface ornamentation.
- C. Lobulate periphery.
- 3) Explain in details the factors controlling the distribution of foraminifera. (20 Marks)

4) Choose the correct answer of the following questions:

(15 Marks)

- 1. Microfossils are generally excellent indicators of
 - a) Tectonics
- b) Earthquake
- c) paleoecology
- d) Paleogeography

- 2. Foraminifera is

 - a) Unicellular animal b) Unicellular plant c) Multicellular animal d) Multicellular plant

- 3. Agglutinated foraminiferal test is formed of
 - a) Calcareous wall
- b) Siliceous wall
- c) Chitineous Walls d) coarse/fine cemented particles
- 4. Porcelaneous foraminiferal test is:
 - a) Perforate
- b) semiperforate
- c) imperforate
- d) non-perforate

- 5. Unilocular foraminiferal test is
 - a) septate
- b) non septate
- c) simply septate
- d) limbate

Best wishes

Examiners	Prof. Mahmoud Faris Mohamed	Prof. Abdelfattah Ali Zalat
	Prof. Akmal Marzouk	